Charles and S

In Class Examples:

Mole-Mass Conversions

Ex 6: How many grams are in 7.20 mol of dinitrogen trioxide?

Prob 18b: What is the mass in grams of 2.40 mol of nitrogen gas?

You do prob 18c: How many grams are in 10.0 mol of chromium metal?

July

Ex 7: Find the number of moles in 92.2 g of iron (III) oxide.
Prob 19b: Find the number of moles in 2.64 x 10 ⁻⁴ a of lithium hydrogen phosphoto
gor numum nyurogen phosphiaic.
Vous do manh 10st Find the munitime Sand 10st 10st
You do prob 19g: Find the number of moles in 187 g Al.